

SUBJECT OUTLINE



Subject Name:

Chemistry and Biochemistry

Subject Code:

BIOB111

SECTION 1 - GENERAL INFORMATION

Award/s:	Total Course Credit Points:	Level:
Bachelor of Health Science (Naturopathy)	128	Core 1 st Year
Bachelor of Health Science (Nutritional and Dietetic Medicine)	96	Core 1 st Year
Bachelor of Health Science (Myotherapy)	96	Core 1 st Year
Bachelor of Complementary Medicine	48	Elective 3 rd Year

Duration: 1 Semester

Subject Coordinator: Dr Eric Owusu Sekyere (Sydney campus)

Subject is: Core or Elective as noted **Subject Credit Points:** 4

Student Workload:

No. timetabled hours per week:	No. personal study hours per week:	Total hours per week:
6	4	10

Delivery Mode:

Face to Face (On campus)	2 x 2 hour lectures	2 x 1 hour tutorials
e-Learning (Online)	Narrated PowerPoint presentations Tutorials: Asynchronous tutor moderated discussion forum and activities Student handouts, web-based resources	
Intensive Delivery (Summer School)	Contact hours are delivered over 4 weeks with 3 x 6.5 hour days delivered per week Content: Combination lecture and tutorial activities Assessment: Online Quiz 1 - Week 1; Online Quiz 2 - Week 2; Written Assignment - Week 4; Final Written Exam - Week 5 Full Time Part Time	
Pre-requisites:	Nil	
Co-requisites:	Nil	

SECTION 2 – ACADEMIC DETAILS

Subject Rationale

The first part of this subject introduces the student to basic and organic chemistry and explores the nature and reactivity of matter. This provides the foundation for the second part – biochemistry – which examines the relationship between the structure and function of complex biomolecules. Students will study the role of enzymes, coenzymes and cofactors in energy metabolism, and metabolic pathways involving glucose, fatty acid and amino acid – providing knowledge of the metabolic processes that occur in the human body. This is a foundational subject for later study of nutrition, pharmacology, immunology, herbal medicine and clinical sciences.

Learning Outcomes

1. Describe elements, atoms, ions, chemical bonding, chemical reactions, the significance of energy transfer and catalysts in biochemical systems.
2. Define the nature of matter and the different states of matter (solid, liquid, and gas) and describe how movement of atoms in the different states impacts their properties.
3. Describe solutions, mixtures and their equilibrium and relate it to how they impact physiological processes and metabolism.
4. Identify and apply nomenclature and common chemical reactions associated with organic functional groups.
5. Describe the components that make up the nucleic acids in cells, describe DNA and RNA structure and alterations in DNA (mutations).
6. Illustrate and analyse the chemical nature of major biochemical groups including carbohydrates, lipids and proteins.
7. Discuss the major biochemical pathways, including metabolism of carbohydrates, lipids and proteins.
8. Investigate the major biochemical pathways in the context of nutrition and health.

Assessment Tasks

Type	Learning Outcomes Assessed	Session Content Delivered	Due	Weighting
Online Quiz 1 multiple choice (40 minutes)	1-2	1-6	Week 4	20%
Online Quiz 2 multiple choice (45 minutes)	3-4	7-15	Week 9	20%
Written Assignment (1200 words)	5-8	7-25	Week 12	20%
Final Written Exam multiple choice, short answers, definitions and extended responses (2.5 hours)	5-7	15-25	Final Examination Period	40%

All written assessments and online quizzes are due at 11:55 p.m. and submitted through the LMS

Prescribed Readings:

1. Stoker, H. S. (2016). *General, organic, and biological chemistry* (7th ed.). Boston, MA: Cengage Learning.
2. Summers, J., & Smith, B. (2014). *Communication skills handbook* (4th ed.). Milton, QLD: Wiley. [ebook available]

Recommended Readings:

1. Alberts, B., Johnson, A., Lewis, J., Raff, M., Roberts, K., & Walter, P. (2008). *Molecular biology of the cell* (5th ed.). New York, NY: Garland Science.
2. Berg, J. M., Tymoczko, J. L., & Stryer, L. (2015). *Biochemistry* (8th ed.). New York, NY: W. H. Freeman.
3. Bettelheim, F. A., Brown, W. H., Campbell, M. K., Farrell, S. O., & Torres, O. J. (2016). *Introduction to general, organic and biochemistry* (11th ed.). Boston, MA: Cengage Learning.
4. Dominiczak, M. H. (2007). *Flesh and bones of metabolism*. Edinburgh, Scotland: Elsevier Mosby.
5. Timberlake, K. C. (2015). *General, organic, and biological chemistry: Structures of life* (5th ed.). Boston, MA: Pearson

6. Tortora, G. J., Derrickson, B., Burkett, B., Peoples, G., Dye, D., Cooke, J., ... Mellifont, R. (2019). *Principles of anatomy and physiology* (2nd Asia-Pacific ed.). Milton, QLD: Wiley.

Subject Content		
Week	Lectures	Tutorials
1.	Session 1 Introduction (Subject Outline / Subject Aims / Assessment / Teaching Resources) Introduction to Chemistry <ul style="list-style-type: none"> Matter and the structure of the atom Periodic table of elements Use of the periodic table to predict physical and chemical properties of elements Electronic configuration and the octet rule Formation of ions Significance of isotopes 	<ul style="list-style-type: none"> Review of atom/subatomic particles, periodic table, and electronic configuration Animations on electron configuration
	Session 2 Chemical Bonding <ul style="list-style-type: none"> Ionic and covalent bonding Polyatomic ions Electro negativity and polarity of bonds Naming ionic and molecular compounds 	<ul style="list-style-type: none"> Review of writing and naming compounds Formative quiz 1
2.	Session 3 Chemical Reactions <ul style="list-style-type: none"> Chemical change Chemical equations Mole Formula weight Balancing equations Classification of chemical reactions Heat of reaction 	<ul style="list-style-type: none"> Review of balancing equations
	Session 4 Reaction Rates <ul style="list-style-type: none"> Le Châtelier's principle and equilibrium 	<ul style="list-style-type: none"> Review of factors affecting chemical equilibrium Formative quiz 2
3.	Session 5 Physical States of Matter <ul style="list-style-type: none"> Solids, liquids and gases Boyle's law Dalton's law Transition between states of matter and intermolecular forces Solubility Kinetic molecular theory 	<ul style="list-style-type: none"> Review of different states of matter, gas laws, types of solutions, solubility, osmosis and dialysis Animations on different states of matter
	Session 6 Acids and Bases <ul style="list-style-type: none"> Reactions of acids and bases, pH and buffers 	<ul style="list-style-type: none"> Review of acids and bases, pH and buffer Virtual labs on the use of indicators in distinguishing acids and bases Formative quiz 3

4.	Session 7 Introduction to Organic Chemistry <ul style="list-style-type: none"> Organic compounds Hydrocarbons Functional groups Stereo isomers 	<ul style="list-style-type: none"> Review of hydrocarbon compounds
	Session 8 Properties of the Functional Groups <ul style="list-style-type: none"> Alkanes Alkenes Alkynes Aromatics Alcohols 	<ul style="list-style-type: none"> Review of alkanes, alkenes, alkynes, aromatics and alcohols Formative quiz 4
5.	Session 9 Properties of the Functional Groups (Continued) <ul style="list-style-type: none"> Phenols Ketones Aldehydes Carboxylic acids Esters 	<ul style="list-style-type: none"> Review of phenols, ketones, aldehydes, carboxylic acids and esters
	Session 10 Properties of the Functional Groups (Continued) <ul style="list-style-type: none"> Ethers Thiols Amines and amides 	<ul style="list-style-type: none"> Review of ethers, thiols, amines and amides
6.	Session 11 Carbohydrates <ul style="list-style-type: none"> An introduction to classification, structure and function of carbohydrates 	<ul style="list-style-type: none"> Concept maps on carbohydrate types
	Session 12 Lipids <ul style="list-style-type: none"> An introduction to classification, structure and function of lipids 	<ul style="list-style-type: none"> Concept maps on lipid types
7.	Session 13 Molecular Biology: Nucleic Acids, Nucleotides, DNA, RNA <ul style="list-style-type: none"> Classification, structure, nucleosides and nucleotides 	<ul style="list-style-type: none"> Review of types of nucleic acids, DNA replication, review DVD on DNA structure
	Session 14 Amino Acids and Proteins <ul style="list-style-type: none"> Introduction to classification, structure and function of amino acids Introduction to protein classification, structure and function 	<ul style="list-style-type: none"> Review of amino acids and proteins Animations on the structure and formation of different levels of protein organisation
NON-TEACHING WEEK (note that make-up classes may be scheduled in this week) Semester 1 - This aligns with the week after Easter so it may fall between Weeks 6 to 8 Semester 2 & Online students - The non-teaching week falls between Weeks 7 and 8		

8.	Session 15 Enzymes and Co-enzymes <ul style="list-style-type: none"> The different classes of enzymes are studied and the types of reactions they catalyse 	<ul style="list-style-type: none"> Review of enzymes Virtual lab on the effect of various factors on enzyme activity
	Session 16 In Class Study Revision	
9.	Session 17 Bioenergy Production <ul style="list-style-type: none"> An overview of the role Integration of the common catabolic pathways 	<ul style="list-style-type: none"> Review of cell structure, role of ATP and co-enzymes, significance of ATP
	Session 18 Metabolism <ul style="list-style-type: none"> Digestion of carbohydrates Glycolysis, pyruvate pathways Glycogen metabolism 	<ul style="list-style-type: none"> Review of carbohydrate metabolism
10.	Session 19 Metabolism (Continued) <ul style="list-style-type: none"> The citric acid cycle The Cori cycle 	<ul style="list-style-type: none"> Concept map on carbohydrate metabolism
	Session 20 Metabolism (Continued) <ul style="list-style-type: none"> Electron transport chain Oxidative phosphorylation 	<ul style="list-style-type: none"> Review of electron transport chain/oxidative phosphorylation Animations of ATP synthesis
11.	Session 21 Metabolism (Continued) <ul style="list-style-type: none"> Gluconeogenesis Hormonal control of carbohydrate metabolism 	<ul style="list-style-type: none"> Review of carbohydrate metabolism
	Session 22 Lipid Metabolism <ul style="list-style-type: none"> β-oxidation, ketogenesis 	<ul style="list-style-type: none"> Review of metabolism of lipids
12.	Session 23 Lipid Metabolism (Continued) <ul style="list-style-type: none"> Fatty acid synthesis 	<ul style="list-style-type: none"> Review of metabolism of lipids
	Session 24 Protein Metabolism <ul style="list-style-type: none"> Urea cycle and synthesis of amino acids 	<ul style="list-style-type: none"> Review of metabolism Concept map on protein metabolism
13.	Session 25 Metabolism (Continued) <ul style="list-style-type: none"> Integrating the metabolic pathways 	<ul style="list-style-type: none"> Concept map on integration of carbohydrates, protein and fat metabolism
	Session 26 In Class Study Revision	
14.	Non-Teaching Week/Practical Examination Week 1 Note that make-up classes may be scheduled in this week	
15.	Non-Teaching Week/Practical Examination Week 2 Note that make-up classes may be scheduled in this week	

<p>16.</p>	<p>Final Examination Week 1</p> <p>On campus enrolled students: Refer to the Examination Timetable for your local campus for the exact day and time of exam</p> <p>Online enrolled students: You are required to sit examinations on campus per the Examination Policy - Higher Education. The Examination Weeks for subjects offered online are identified in the Online Calendar</p>
<p>17.</p>	<p>Final Examination Week 2</p> <p>On campus enrolled students: Refer to the Examination Timetable for your local campus for the exact day and time of exam</p> <p>Online enrolled students: You are required to sit examinations on campus per the Examination Policy - Higher Education. The Examination Weeks for subjects offered online are identified in the Online Calendar</p>